

# CHEMISTRY (CHE)

## Courses and Descriptions

### CHE 100 Intro to College Chemistry 3 Credits

Open to all students, but designed primarily for those who wish to major in a science which requires chemistry but whose chemistry background is not sufficient to allow entrance into Chemistry 120. It focuses on the nomenclature used in chemistry including the symbols used to designate the chemical elements, the construction of chemical formulas, and the writing and balancing of chemical equations. Other topics will include interpreting the Periodic Table, the valences of the elements, the mole concept, and simple stoichiometry. In addition, chemical calculations involving units, scientific notation, significant figures, and the algebraic manipulations of simple equations will be included. Three hours of lecture per week. This course does not satisfy the requirements for the biochemistry or chemistry degree, but does satisfy the core requirements for education and business majors.

### CHE 110 Survey of General Chemistry 4 Credits

A one-semester survey of basic general chemical principles designed to be applied to questions in health-related fields. Students will explore the fundamental structures of atoms and simple compounds that comprise living beings; basic patterns in chemical reactivity, especially oxidation-reduction and acid-base reactions; quantitative analyses of biologically-relevant compounds and chemical reactions; properties of substances; and kinetic and thermodynamic principles underlying such chemicals and chemical processes. In the laboratory portion of this course, students will learn and apply principles of experimental safety and will apply knowledge gained in the class via application in basic experimental contexts. No prerequisites. Students who have completed CHE 120 and/or CHE 122 will not be permitted to enroll in CHE110/110L. Furthermore CHE 110 will not serve as a prerequisite for CHE 122.

### CHE 110L Survey of Gen Chemistry Lab 0 Credits

In the laboratory portion of this course, students will learn and apply principles of experimental safety and will apply knowledge gained in the class via application in basic experimental contexts.

### CHE 114 Chemistry in the Kitchen 4 Credits

Chemistry permeates aspects of our daily lives in which we are often unaware. In this course, students will learn the core tenets of chemistry including atomic and molecular structure, bonding, intermolecular and macromolecular interactions, and chemical reactivity, and will personally investigate these properties in the context of cooking, baking, metabolism, and other kitchen-related activities. This 4-credit course will include a weekly 3-hour lab in which students will perform experiments that allow them to prepare dishes that illustrate key chemical concepts. As part of the course, students will collaborate with the Trenton Area Soup Kitchen to prepare and serve food to the community, as well as share their understanding about the chemical properties that are involved in the development of various dishes. This course counts towards the fulfillment of the Disciplinary Perspectives element of the CLAS general education curriculum.

**Corequisite(s):** CHE 114L.

### CHE 114L Chemistry in the Kitchen Lab 0 Credits

This is the laboratory portion of CHE 114.

**Corequisite(s):** CHE 114.

### CHE 115 Chemistry & Society 3 Credits

Designed to give the nonscientist an appreciation of the role of chemistry in today's world. The approach is conceptual rather than mathematical. Topics include basic principles of chemical theory, energy sources, elementary organic chemistry, drugs, food additives, polymers, chemistry of living systems, inorganic solids in modern technology, and problems involving pollution of the environment. Three hours of lecture per week. This course satisfies the core requirements for education and business majors.

### CHE 118 Exploration of Chemical Principles 4 Credits

A one-semester introduction to the principles of chemical sciences. Students will utilize inquiry-based learning methods to examine contextual problems as a means to explore introductory models and concepts of chemistry. Students will also gain an understanding of how scientific models are used to explain experimental observations. The laboratory component of this course is designed to provide students with an experimental context within which to develop some of the models described in the classroom. Three hours of lecture and one three-hour lab per week.

### CHE 118L Exploration of Chemical Principles Lab 0 Credits

This lab is a co-requisite and must be taken with the corresponding course.

**Corequisite(s):** CHE 118.

### CHE 120 Principles of Chemistry 3 Credits

For students who have successfully completed one year of high school chemistry. This systematic study of the fundamental principles and concepts of chemistry covers atomic structure, bonding, stoichiometric relationships, including solution and oxidation-reduction reactions, and molecular structure. Three hours of lecture per week. Prerequisite(s): High school chemistry or CHE 100 is recommended before taking this course.

**Corequisite(s):** CHE 121.

### CHE 121 Principles of Chemistry Lab 1 Credits

For students concurrently taking CHE 120. Experiments involve gravimetric, volumetric, and spectrophotometric quantitative analysis. One three-hour lab per week. Fall.

### CHE 122 Intro to Chemical Systems 3 Credits

A continuation of CHE 120. For students majoring in the sciences but may be taken by others. Chemical systems in which the study of kinetics, thermodynamics, equilibrium, and radiochemistry are emphasized. Three hours of lecture per week. Prerequisite(s): CHE 120, MTH 105 or higher.

**Corequisite(s):** CHE 123.

### CHE 123 Quantitative Methods Lab 1 Credits

Usually taken concurrently with CHE 122. Primarily for students majoring in the sciences. A number of quantitative classical and instrumental methods of analysis are used to determine thermodynamic properties and reaction mechanisms. One three-hour lab per week. Prerequisite(s): CHE 121.

**Corequisite(s):** CHE 122.

**CHE 211 Organic Chemistry I 4 Credits**

The structure, chemical properties, and methods of preparation of the more important classes of carbon compounds are studied, with an emphasis on the relationship of structure, stereochemistry, and conformation to chemical reactivity. The preparation and reactivity of organic functional groups is introduced. The use of infrared and nuclear magnetic resonance spectroscopy, and mass spectrometry for elucidating structures of organic molecules is discussed. Three hours of lecture and one three-hour lab per week. Prerequisite(s): CHE 120, CHE 121.

**Corequisite(s):** CHE 211L.

**CHE 211L Organic Chemistry I Lab 0 Credits**

This lab is a co-requisite and must be taken with the corresponding course.

**Corequisite(s):** CHE 211.

**CHE 214 Organic Chemistry II 4 Credits**

A continuation of Chemistry 211, emphasizing the mechanism of organic reactions, structural interpretations of properties, preparations, reactivity and identification of organic compounds. Three hours of lecture and one three-hour lab per week.

**Prerequisite(s):** CHE 211 and CHE 211L.

**CHE 214L Organic Chemistry II Lab 0 Credits**

This lab is a co-requisite and must be taken with the corresponding course.

**Corequisite(s):** CHE 214.

**CHE 225 Introduction to Organic and Biochemistry 4 Credits**

An introductory course describing the basic principles of organic chemistry and biochemistry as they relate to human metabolism and disease. The nature of the chemical structure and reactivity of organic functional groups such as alcohols, aldehydes, ketones, carboxylic acids and amines will be presented with biological processes in mind. The biochemistry of the macromolecules DNA, RNA, proteins, carbohydrates and lipids will be discussed leading in to a discussion of some of the more important metabolic pathways. This course is intended for science majors who do not take the full two semester sequence of organic chemistry and two semesters of biochemistry and desire a background in biochemistry. Non-science major students who have had one semester of general chemistry and one semester of biology may also enroll in the course. Three hours of lecture and one three-hour lab per week.

**Prerequisite(s):** ( CHE 110 UG D and CHE 110L ) or ( CHE 122 UG D and CHE 123 ) and BIO 115 .

**Corequisite(s):** CHE 225L.

**CHE 225L Introduction to Organic and Biochemistry Lab 0 Credits**

This lab is a co-requisite and must be taken with the corresponding course.

**Prerequisite(s):** (CHE 110 with a minimum grade of D and CHE 110L) or (CHE 122 with a minimum grade of D and CHE 123) and BIO 115

**Corequisite(s):** CHE 225.

**CHE 250 Quantitative Analysis and Statistics Methods 4 Credits**

This course will provide a deeper exploration of topics in chemistry that are steeped in numerical analysis. These topics include advanced analysis of equilibrium systems, acid-base systems and electrochemical systems. Additional detail will be given to methods of chemical measurement, statistical methods of data analysis and determination of data validity and reliability. Both lecture and laboratory will show an emphasis on using computer-based tools to analyze experimental data. Three hours of lecture and one three-hour lab per week. Prerequisite(s): CHE 122 & CHE 123

**Corequisite(s):** CHE 250L.

**CHE 250L Quantitative Analysis and Statistical Methods Lab 0 Credits**

This lab is a co-requisite and must be taken with the corresponding course.

**Corequisite(s):** CHE 250.

**CHE 305 Physical Chemistry I 3 Credits**

The mathematic and conceptual foundations of physical chemistry will be introduced with an over-arching theme of determination of energy allocation within atomic and molecular systems. Topics will include determination and measurement of energy states in atoms and molecules, simple quantum mechanical systems, distribution of energies and the connection to thermodynamic quantities, the three laws of thermodynamics, spontaneity, equilibrium and experimental kinetics.

**Prerequisite(s):** a minimum grade of C- in CHE 122, CHE 214, MTH 211, and (PHY 101 or PHY 201).

**CHE 306 Physical Chemistry II 3 Credits**

Physical chemistry concepts are explored in more detail with emphasis on examination of systems that require multiple models in physical chemistry to explain. Topics will include, kinetic theory and transition state theory, statistical mechanics and its connections to thermodynamic functions, temperature dependence of spontaneity and equilibrium, the thermodynamics of condensed phases and multi-component equilibria, electrochemistry, multi-electron quantum mechanical systems, approximations in quantum mechanics, symmetry and advanced molecular spectroscopy.

**Prerequisite(s):** CHE 305, MTH 211.

**CHE 315 Inorganic Chemistry 3 Credits**

The periodic table as a tool for predicting the physical and chemical properties of chemical systems is developed and examined in conjunction with various theories of bonding, including valence bond, molecular orbital, valence shell electron repulsion, and ligand field theory. Emphasizes structure of crystalline solids, coordination compounds, reaction mechanisms, and structure-property relationships. Three hours of lecture per week.

**Prerequisite(s):** CHE 122, CHE 214.

**CHE 316 Inorganic Chemistry Laboratory 1 Credits**

Students will explore a variety of synthetic methodologies for the growth of inorganic molecular systems, and solid-state materials. A transition metal or main group metal plays a central structural role in all systems that will be examined. Modern analytical methods will be applied to characterize synthesis products, such as FT-IR spectroscopy, polarimetry, NMR, and powder X-ray diffraction analysis. Prerequisite(s): CHE 123, CHE 214.

**Corequisite:** CHE 315.

**CHE 320 Polymer Chemistry 3 Credits**

Designed to acquaint students with the structure and properties of polymers, the contrast between small molecules and polymers, methods of measuring molecular weight, the mechanism of polymerization, and the methods of fabricating polymers.

**Prerequisite(s):** CHE 211.

**CHE 325 Physical Chemistry Laboratory 1 Credits**

Measurements are made of physical properties of molecules and chemical dynamical processes. These measurements will be used to develop models which explain the physical chemical nature of the systems under examination. Experiments will utilize various instrumental techniques such as infrared spectrometry, nuclear magnetic resonance, fluorescence and UV/Vis spectrometry. One three-hour lab per week.

**Prerequisite(s):** CHE 250, CHE 305.

**CHE 330 Instrumental Analysis Lecture 4 Credits**

This course is designed to provide students with background theory and principles of operation of modern instrumental methods. The instrumental methods in this course include, but are not limited to, ultraviolet and visible spectroscopy, atomic spectroscopy, gas chromatography (GC), high performance liquid chromatography (HPLC), capillary electrophoresis, mass spectrometry (MS) and electrochemical analysis.

**Prerequisite(s):** a minimum grade of C- in CHE 214, CHE 250, and (PHY 101 or 201).

**CHE 330L Instrumental Analysis Laboratory 0 Credits**

This course is designed to give students practical experience using modern analytical instrumentation. The instrumental methods in this course include, but are not limited to, ultraviolet and visible spectroscopy, atomic spectroscopy, gas chromatography (GC), high performance liquid chromatography (HPLC), capillary electrophoresis, mass spectrometry (MS) and electrochemical analysis.

**Prerequisite(s):** a minimum grade of C- in CHE 214, CHE 250, and (PHY 101 or PHY 201).

**CHE 340 Environmental Chemistry 3 Credits**

This course is an advanced quantitative and qualitative study of the sources of environmental pollutants, their chemical reactions and mechanism of formation, their fate in the environment, and ultimately their impact on environmental systems and services including ozone cycling, ozone hole formation, smog formation, heavy metals in drinking water, energy, and global climate change. This course will review scientific literature to identify advances in science and technology to combat these environmental problems. The course is aimed principally at chemistry majors, but other science majors that desire a chemical understanding of the environment and meet the course prerequisites are welcome.

This course satisfies the global perspectives (GP) requirement at Rider University. Recommended courses include MTH 210 and CHE 250.

**Prerequisite(s):** (BCH 225 and 225L) or (CHE 211 and 211L) or (CHE 225 and CHE 225L) and MTH 105 or higher.

**CHE 350 Advanced Organic Synthesis 4 Credits**

Students explore modern topics in organic synthesis including stereoselective synthesis, photocatalysis, and organometallic cross-coupling and metathesis. The laboratory is an authentic research experience wherein students learn and exploit advanced methods in organic synthesis and characterization including Schlenk and glovebox techniques, computer-assisted retrosynthesis and property prediction, and two-dimensional NMR spectroscopy, while considering green chemistry principles throughout.

**Prerequisite(s):** a minimum grade of C- in CHE 214.

**CHE 350L Advanced Organic Synthesis Laboratory 0 Credits**

Students explore modern topics in organic synthesis including stereoselective synthesis, photocatalysis, and organometallic cross-coupling and metathesis. The laboratory is an authentic research experience wherein students learn and exploit advanced methods in organic synthesis and characterization including Schlenk and glovebox techniques, computer-assisted retrosynthesis and property prediction, and two-dimensional NMR spectroscopy, while considering green chemistry principles throughout.

**Prerequisite(s):** a minimum grade of C- in CHE 214.

**CHE 375 Computational Chemistry Lab 2 Credits**

This course will provide students with a means to explore various methods in computational chemistry. Basic computational methods will be developed from first principles and these methods will then be tested using various modeling and computational software packages. Methods will include, but are not limited to, molecular mechanics, semi-empirical molecular orbital calculations, ab initio methods, and density functional calculations. Students will be exposed to various computational software packages and the strengths and limitations of each methodology will be explored. The course will meet in a computer laboratory for two for three-hour lab periods a week.

**Prerequisite(s):** CHE 305.

**CHE 400 Chemical Bonding 3 Credits**

The effects of the chemical bond on the structure and properties of molecules are investigated. Molecular orbital theories of bonding are introduced. Emphasis is placed on group theoretical methods utilizing molecular symmetry to simplify the description of the electronic structure of molecules and to predict their geometric structures and reactivity.

Three hours of lecture per week.

**Prerequisite(s):** CHE 305, MTH 212.

**CHE 415 Special Topics in Chemistry 3 Credits**

An advanced level of one or more areas of modern chemistry. Emphasis on research and the literature of an area of current importance. Three hours of lecture per week.

**CHE 420 Physical Organic Chemistry 4 Credits**

In-depth studies of the methods for elucidating mechanisms of organic reactions for students who have completed one year of organic chemistry and physical chemistry. Topics include conformational analysis, linear free energy relationships, frontier molecular orbital theory, transition state theory, and chemical reaction kinetics. Isotopic scrambling, kinetic isotope effects, NMR and IR spectroscopy, polarimetry, and ultraviolet-visible spectrophotometry will be employed to investigate these concepts. Three hours of lecture and one three-hour lab per week.

**Prerequisite(s):** CHE 214, CHE 305.

**Corequisite(s):** CHE 214L.

**CHE 420L Physical Organic Chemistry Lab 0 Credits**

This lab is a co-requisite and must be taken with the corresponding course.

**Corequisite(s):** CHE 420.

**CHE 490 Independent Study: Research and Creative Expression 1-4 Credits**

Immerses the student in laboratory research. The student learns to organize material, use the literature, make precise measurements, and obtain reproducible data. If possible, the student will publish the results or present them at a scientific meeting.

**CHE 491 Internship in Chemistry 1-4 Credits**

A supervised research experience in an approved organization where qualified students gain real-world knowledge and utilize their academic training in a professional environment. Placement may be in a private, public, non-profit, or governmental organizations under the guidance of a mentor. The mentor and student will have regular consultation with the departmental internship coordinator to assess the student's progress. Normally, 50 hours of internship per credit is required. The grade for the course will be determined by the students' overall performance in their research work, a research paper documenting their work with their internship mentor and an oral or poster presentation at the end of the semester.

**Prerequisite(s):** 2.5 GPA and permission of the instructor.